

J.T.Baker BAKERBOND Polymeric Multimode Ion Exchange Chromatography Media

Ion exchange chromatography is one of the most widely used methods in purification of biopharmaceutical drug such as recombinant protein, monoclonal antibody, antibody related product, and DNA.

This documents contains technical information of BAKERBOND™ PolyIEX media and column packing procedure.

BAKERBOND™ PolyIEX media are based on spherical polymethacrylate beads covalently bonded to polyethylenimine and additional ionic groups to offer primary anion or cation functionalities along with secondary anion exchange sites.

This unique design results in multimodal behavior and offers enhanced selectivity and high efficiency when compared to conventional ion exchange media. PolyIEX is alkaline stable and its operational flexibility makes it ideal for use in various chromatographic purifications, where PolyIEX can be packed with relative ease. PolyIEX offers the following attributes to provide the highest resolution and productivity:

- Unique selectivity
- High efficiency
- Scalable and robustness
- Manufactured in GMP with appropriate quality system
- Global application support

| Description | Cat No. |
|-------------|---------|
| Poly PEI | 7585 |
| Poly Quat | 7603 |
| Poly ABx | 7586 |
| Poly CSx | 7587 |



Information

| Property | PolyPEI | PolyABx | PolyCSx | PolyQUAT |
|-----------------------|--|--|--|--|
| Functionality | Primary weak anion exchanger + hydrophilicity | Multimode Primary weak cation exchanger with weak anion exchange sites+ hydrophilicity | Multimode Primary strong cation exchanger with weak cation and anion exchange sites + hydrophilicity | Multimode Primary strong anion exchanger with weak anion exchange sites + hydrophilicity |
| Functional Group | CH2CH2NH | -NH-C(=0)-CH2-CH2- COOH | -NH-C(=0)-CH(SO3H)- CH2-COOH | -NH-CH2-CH(OH)-CH2- N+(CH3)3 |
| Ionic Capacity | 0.3–0.5 meq/mL as chloride | 0.15–0.25 mmol H+/mL (dynamic) | 0.3-0.5 meq/mL | 0.4-0.6 meq/mL |
| Average Particle Size | 35 μm | | | |
| Median Pore Diameter | 500 Å | | | |
| Porosity | 4 x 106 (globular proteins) | | | |
| Operating pH Range | 4.5–9 | 4.5-14 | 4.5-14 | 4.5-14 |
| Cleaning (CIP) | 50 mM Phosphoric acid ; 0.1-1.0 N NaOH; pH range: 1.0-14 | | | |
| Chemical Stability | All commonly used aqueous buffers, sodium hydroxide, acetic acid, phosphoric acid, urea, guanidine hydrochloride, up to 100% ethanol, methanol, or isopropanol | | | |

Stability

| | PolyPEI | PolyABx | PolyCSx | PolyQUAT |
|--------------------------------------|--|---------|---------|---------------------|
| Storage Stability of Media / C | Storage Stability of Media / Column | | | |
| Storage Buffer* | 5 yr | 5 yr | 5 yr | 5 yr |
| H₃PO₄ 50 mM | 5 yr | 5 yr | 5 yr | 5 yr |
| NaOH 0.1 M | 5 yr | 5 yr | 5 yr | 5 yr |
| Clean-in-Place Stability of Me | Clean-in-Place Stability of Media / Column | | | |
| H ₃ PO ₄ 50 mM | > 48 h | > 48 h | > 48 h | > 48 h |
| NaOH 0.1 M | > 48 h | > 48 h | > 48 h | > 48 h |
| NaOH 0.5 M | > 2 h 0.5 h, 80℃ | > 2 h | > 2 h | > 2 h 0.5 h, 80℃ |
| Reuse Stability | | | | |
| Column Pack/Unpack | | > 100 | | |

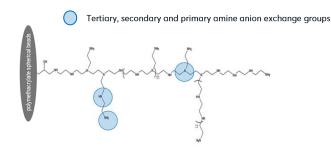
Media are also stable in ethanol or GuHClStability is indicated by capacity, separation and particle size

^{*} Storage buffer : 200 mM Sodium acetate, 2% benzyl alcohol, pH 4.2

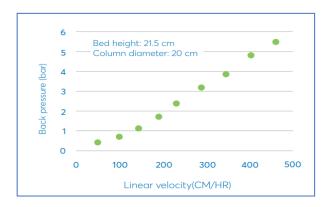
JT. Baker BAKERBOND Poly PEI

Weak anion exchanger chromatography media

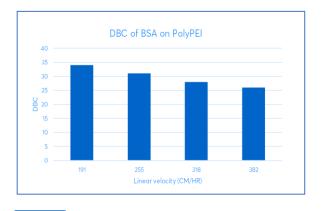
BAKERBOND PolyPEI is a multimode ion exchanger functioning as a weak anion exchanger over a wide pH range. The unique selectivity of PolyPEI is the result of proprietary surface chemistry with weak anionic exchange sites due to the presence of primary, secondary and tertiary nitrogen on the Polyethyleneimine (PEI) ligands. The presence of amino groups with different pKa offers better selectivity than conventional weak anion exchange media with equivalent capacities, and it is capable of separating proteins and peptides having similar isoelectric points(pI). Unique selectivity of closely related molecules is often achieved with PolyPEI where conventional ion exchangers fail to provide sufficient separation in a process environment.



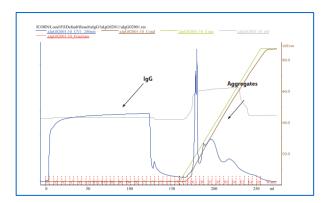
| Property | PolyPEI | |
|---------------------|--|--|
| Functionality | Primary weak anion exchanger + hydrophilicity | |
| Functional Group | CH2CH2NH | |
| Ionic Capacity | 0.3–0.5 meq/mL as chloride | |







Dynamic Binding Capacity of BSA on PolyPEI at High Linear Velocity



IgG Flow-through mode application

Mode: Flow-through application

Sample: hlgG

Resin: BAKERBOND PolyPEI EQ Buffer: 50 mM MES pH 6.5

Elution Buffer : 50 mM MES, 1 M NaCl, pH 6.5 Linear gradient from 0 to 100% B in 10 CV

Flow rate: 1.2 mL/minute

Flow-through mode with PEI is applied for hlgG $\,$

purification.

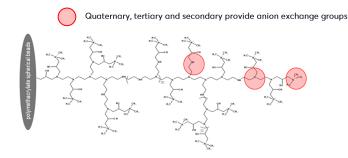
Aggregate form and other impurities are removed during

this purification. Purified IgG is collected by F/T.

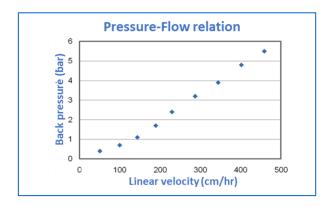
J.T.BakerPolyQuat

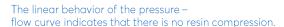
Strong anion exchanger chromatography media

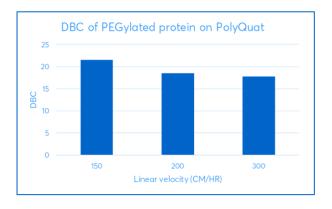
BAKERBOND PolyQuat is a multimode ion exchanger functioning as a strong anion exchanger over a wide pH range. Secondary anion exchange sites are due to the presence of amine groups. This multimodal functionality offers better selectivity than conventional strong anion exchange media with equivalent capacities and is capable of separating proteins and peptides having similar isoelectric points (pl). Unique selectivity of closely related molecules is often achieved with PolyQUAT where conventional ion exchangers fail to provide sufficient separation in a process environment.



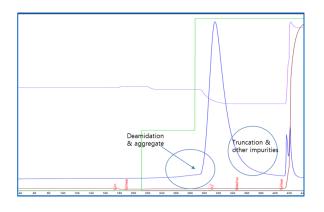
| Property | PolyQuat | |
|---------------------|---|--|
| Functionality | Multimode Primary strong anion exchanger with weak anion exchange sites + hydrophilicity | |
| Functional Group | -NH-CH2-CH(OH)-CH2-N+(CH3)3 | |
| lonic Capacity | 0.4-0.6 meq/mL | |







Dynamic Binding Capacity of BSA on PolyQuat High Linear Velocity



Protein polishing step application

Resin: BAKERBOND PolyQuat

Sample: rProtein Polishing step application EQ Buffer: 50 mM Sodium phosphate, pH 8.0 Elution Buffer: 65 mM Phospho-citrate, pH 5.8

PolyQuat is applied for removing deamidation, truncation and other form of impurity.

Peak is separated 3 parts which are main target drug peak and other 2 impurity peaks.

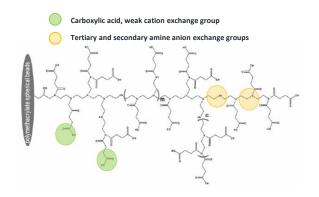
Front of main peak contains deamidation and aggregate.

Tail of main peak contains truncation and other impurities.

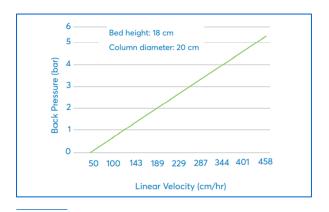
J.T.BakerBAKERBOND Poly ABx

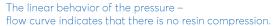
Weak cation exchanger chromatography media

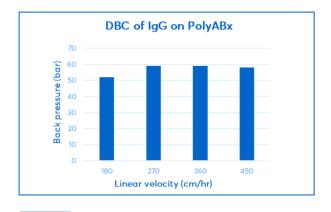
BAKERBOND ABx is a multimode ion exchanger which primarily functions as a weak cation exchanger over a wide pH range. Secondary anion exchange sites are due to the presence of amine groups. This multimodal functionality offers better selectivity than conventional weak cation exchange media with equivalent capacities and is capable of separating proteins and peptides having similar isoelectric points (pl). Unique selectivity of closely related molecules is often achieved with PolyABx where conventional ion exchangers fail to provide sufficient separation in a process environment.



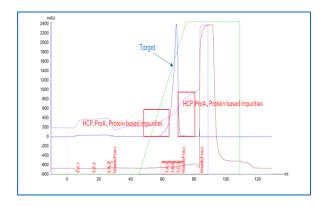
| Property | PolyABx | |
|---------------------|--|--|
| Functionality | Multimode Primary weak cation exchanger with weak anion exchange sites + hydrophilicity | |
| Functional Group | -NH-C(=0)-CH2-CH2-COOH | |
| lonic Capacity | 0.15–0.25 mmol H+/mL (dynamic) | |







Dynamic Binding Capacity of BSA on PolyABx High Linear Velocity



FC fusion protein polishing step application

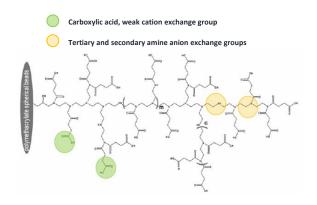
Resin: BAKERBOND PolyABx Sample: Fc fusion protein polishing step application EQ Buffer: 50 mM Phospho-citrate, 5 mM Sodium chloride pH 5.0 Elution Buffer: 50 mM Sodium phosphate, pH 8.0

PolyABx is applied in polishing step of Fc fusion protein purification. HCP, residual protein A and other impurities are removed at this step. In residual protein A, the input contains 15.2 ppm of residual protein A, but the output contains 1.1 ppm (> 90 % of reduction rate).

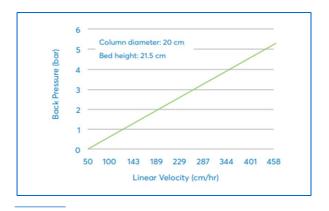
J. T. Baker BAKERBOND Poly CSx

Strong cation exchanger chromatography media

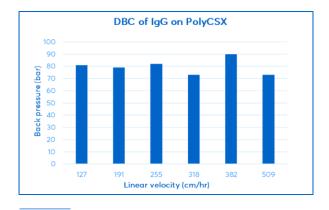
BAKERBOND CSx is a multimode ion exchanger functioning as a strong cation exchanger over a wide pH range. Its strong cationic exchange stites are a result of sulfonic acid group. Weak cationic exchange sites are due to carboxylic acid groups, while weak anion exchange behavior is maintained by the presence of amino groups on the Polyethyleneimine (PEI) ligands. This multimodal functionality offers better selectivity than conventional strong cation exchange media with equivalent capacities and is capable of separating proteins and peptides having similar isoelectric points (pI). Unique selectivity of closely related molecules is often achieved with PolyCSx where conventional ion exchangers fail to provide sufficient separation in a process environment.



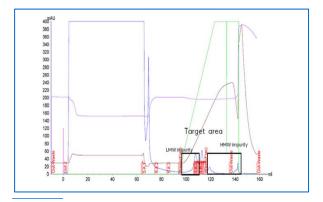
| Property | PolyABx | |
|---------------------|--|--|
| Functionality | Multimode Primary weak cation exchanger with weak anion exchange sites + hydrophilicity | |
| Functional Group | -NH-C(=0)-CH2-CH2-COOH | |
| lonic Capacity | 0.15–0.25 mmol H+/mL (dynamic) | |







Dynamic Binding Capacity of BSA on PolyCSx High Linear Velocity



Fab capture step application

Resin: BAKERBOND PolyCSx Sample: Fab capture step application EQ Buffer: 50 mM Phospho-citrate, 50 mM Sodium chloride pH 6.5 Elution Buffer: 50 mM Phospho-citrate, 500 mM Sodium chloride, pH 6.5

Poly CSx is applied in capture step for Fab purification. The input has 15.70 % purity, but the output has 79.44 % purities in SE-HPLC.

Column packing instruction

Formula

- Compression factor = $\frac{\text{Setled bed height}}{\text{Packed bed height}}$
- Column volume = column bed height (cm) x cross sectional area (cm²)
- Settled media volume = column volumn x compression factor
- Slurry volume = Settled media volume Slurry concentration (%)

Packing guidance

The mobile phase, beforehand, needs to be determined in order to do the column packing on the $\ddot{A}KTA^{TM}$ system. The solvent can be determined as described in the table below.

- 1. Assemble the coloumn in accordance with instruction.
- 2. Connect the column to the stand holder or FPLC holder.
- 3. Mark the target height on the column.
- 4. Fill the column with around 1 cm packing buffer.
- 5. Load the column with calculated resin slurry.
- 6. Start down-flow into column.
- 7. Follow step-wise flow until delta-pressure reaches 90% of max pressure and check bed consolidation.
- 8. Calculate the final bed height.
- 9. The column is ready for the evaluation.

Evaluation

| | Conductivity | UV280 |
|---------------------|-----------------------|---------------------|
| Evaluation sample | 0.8 M Sodium chloride | 1% Acetone solution |
| Evaluation solution | 0.4 M Sodium chloride | Water |
| Injection volume | 1% of column volume | 1~5% |
| Flow rate | 50 cm/hr | 50 cm/hr |

HETP and As are calculated from curve as below.

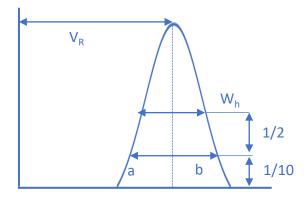
HETP =
$$\frac{L}{N}$$

N = 5.54 × $(\frac{V_R}{W_*})^2$

The reduced plate height is described h and calculated as below. Less than 5 means high efficiency.

$$h = \frac{HETP}{d_{50v}}$$
Asymmetry = $\frac{b}{a}$

The peak has to be symmetrical, and the asymmetry factor has to be as close to 1 as possible and acceptance criteria are from 0.8 to 1.8.



L = bed height

N = Number of theoretical plate

 $V_R = Max peak height volume$

 W_h = Peak width of 50% points

d_{50v} = Median particle size (cm)

A = front peak width at 10% points

B = Tail peak width at 10% points

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